

Corn & Soybean News

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COLLEGE OF AGRICULTURE, FOOD AND ENVIRONMENT Grain and Forage Center of Excellence

Kentucky.

Calcium or Lime? Which Raises Soil pH? A Follow-up.

E arlier in the year we wrote a Corn & Soybean News article (*Calcium or Lime? Which Raises Soil pH?*) discussing the differences between calcium and lime and how they influence soil pH. A study stemmed from several agriculture extension agents and ag retailers receiving questions about a "liquid calcium" (Advanced-Cal, AgriTec International) product that claimed to adjust soil pH using a fraction of the ag lime rate. For example, claims were made that 2 to 4 gallons of Advanced-Cal per acre was more effective at adjusting soil pH than a ton or more of agricultural limestone per acre. The liquid calcium product is calcium chloride (CaCl₂) which has no liming ability (Equation 1).

Equation 1. $CaCl_2 + 2H^+ \rightarrow Ca^{2+} + 2H^+ + 2Cl^-$

To neutralize acidity, the proton (H⁺) must be consumed/neutralized. The neutralization reaction, when calcitic limestone is used, is shown in Equation 2.

Equation 2. CaCO₃ + 2H⁺ \rightarrow H₂CO₃ à H₂O + CO₂ + Ca²⁺

The acidity, H⁺, in Equation 1 remains after the addition of CaCl₂, but in Equation 2 the proton (H⁺) is neutralized to form water by the carbonate present in the limestone. A liming product MUST contain carbonates (CO₃²⁻), hydroxides (OH⁻), or oxides (O²⁻). Carbonate forms (agricultural limestones) are the most common source used in agricultural production. The calcium, magnesium or other ions associated with limestone have nothing to do with the neutralizing reaction.

Field and lab incubation studies were established concurrently in the summer of 2021 and ended approximately one year later. The field study was at 16 locations with 3 reps and the laboratory in-

cubation study, with 4 reps, was done at the UKREC. The same treatments were used for both studies: an untreated check (nothing applied), liquid calcium at 5 gal/acre, pelletized lime (RNV 83), and agricultural lime (RNV 79). Both pelletized and ag lime were applied at 2 ton RNV 100 equivalent lime/ acre, adjusting the rate for the product RNV. Field study soil samples field study were collected prior to treatment application, again approximately 3 months later and again approximately 12 months later. The incubation study used a Crider silt loam soil with an initial pH of 5.2. Soils were placed in 4-oz specimen cups with small holes in the cap to allow for gas exchange. Cups were maintained at 80% water filled pore space by weight until just before the 6-month sample date, when the building was destroyed by the tornado. Samples were recovered but cup moisture wasn't maintained for the 12month samples. Destructive sampling occurred at 1, 3, 6 and 12 months. Results for the field study are reported as the average soil pH across locations and as the change in soil pH due to the sites having different initial soil pH levels. Results for the incubation study are reported as average pH across the reps. Data was analyzed using SAS version 9.4 (Cary, NC).

The chemistry associated with the neutralization reaction in equation 2 was confirmed by the field and laboratory studies (Table 1 and Table 2). Further, the lack of pH change in the liquid calcium and check treatments in these experiments is due to the lack of proton (acidity) neutralization, as indicated in Equation 1. The soil pH 3 months after treatment application had increased with both agricultural and pelletized limestone application, but not in the un-treated check and liquid calcium treatments. The lime materials increased soil pH by 0.30 to 0.43 pH units, while the check and liquid calcium amended soil pH decreased slightly during the first three months (Table 1). The lime materials had increased soil pH by 0.67 to 0.77 units approximately 12 months after treatment applications. The check and liquid calcium treated soils both exhibited slight increases in pH about one year after application, but this was not due to a treatment effect, but rather to an environmental effect. There can be seasonal variation in pH due to soil moisture levels, largely associated with soluble fertilizer salts remaining in the soil. The use of a salt solution to determine pH helps to reduce this variation but does not remove 100% of this variation. It is very apparent that liquid calcium was not effective in raising soil pH as compared to the lime materials. The full neutralization reaction of limestone takes between 2 to 3 years in field settings.

The laboratory incubation results support those found in the field experiment (Table 2). The lime materials increased soil pH after 1 month, relative to the check and liquid calcium treatments. These results were also observed at 3, 6 and 12 months of incubation. Soil moisture was maintained at approximately 80% pore filled volume until shortly before the 6-month sampling, an F4 tornado destroyed the building where the incubating samples were stored. The samples were recovered but their 80% pore filled volume status was not maintained for the rest of the experiment. The drier soil conditions in the 12-month samples could explain why the lime material treatments showed a slight pH decrease over that time period. This does coincide with field observations over the same time period. Soil acidity is neutralized by the consumption of protons, not the addition of calcium. The results of this study support the chemistry in Equations 1 and 2. Liming acid soils should be based on products that contain carbonates, oxides or hydroxides, the associated RNV, and using an application rate based on soil pH and buffer pH - NOT the amount or form of calcium present in a product.

Table 1. Soil	H 0, 3 and 12 Months After Field Application. DataSoil pH			is averaged over 16 field sitesChange in Soil pH	
Treatment	Initial soil pH (<0.854)	Soil pH (3 m) (<0.001)	Soil pH (12 m) (<0.001)	Change in 3 m ² (<0.001)	Change in 12 m ³ (<0.001)
Check	5.76 a ¹	5.64 a	5.87 a	-0.10 a	0.11 a
Liquid Calcium	5.76 a	5.71 a	5.83 a	-0.02 a	0.08 a
Pelletized Lime	5.82 a	6.07 b	6.48 b	0.30 b	0.67 b
Ag Lime	5.76 a	6.16 b	6.51 b	0.43 b	0.77 b

¹ Different letters within a column indicate significant differences at the 0.01 probability level.

² Change after 3 months (soil pH at 3 month – initial soil pH)

³ Change after 12 months (soil pH at 12 months – initial soil pH)

Table 2. Soil pH After 1, 3, 6 and 12 Months of Laboratory Incubation. Initial pH was 5.2.						
	Incubation Time in months and (Pr>F)					
Treatment	1 (<0.001)	3 (<0.001)	6 (<0.001)	12 (0.001)		
Check	5.20 a ¹	5.05 b	4.99 a	5.10 a		
Liquid Calcium	5.25 a	5.00 a	4.98 a	4.98 a		
Pelletized Lime	5.93 b	6.30 d	6.45 b	6.26 b		
Ag Lime	6.10 b	6.20 c	6.38 b	6.20 b		

¹ Different letters within a column indicate significant differences at the 0.01 probability level.



Dr. Edwin Ritchey Extension Soils Specialist (859) 562-1331 edwin.ritchey@uky.edu



Dr. John Grove Professor of Agronomy/ Soils Research and Extension (859) 562-1301 jgrove@uky.edu



Dr. Chris Teutsch Extension Forage Specialist (859) 562-1334 chris.teutsch@uky.edu

University of Kentucky Crop Pest Management Webinar Series continues

he University of Kentucky has once again organized webinars on field crop protection topics that will be hosted through the Southern Integrated Pest Management Center. The weekly webinars began Nov. 8 and feature University of Kentucky Extension Specialists speaking on topics ranging from Weed Science, Plant Pathology and Entomology.



Kentucky Pesticide Applicator and Certified Crop Advisor continuing education credits are available. Preregistration for the webinars is required through the registration URL provided. Dates, speakers and registration links are listed below. All webinars will begin at 10 a.m. EST/ 9 a.m. CST, on the Tuesday morning listed. For more information contact Jason Travis, Agricultural Extension Associate for the University of Kentucky, at (859) 562-2569 or by email at jason.travis@uky.edu.



Date: November 15, 2022

Speaker: Dr. Carl Bradley
Title: Managing Important Soilborne Diseases of Soybean in Kentucky
Registration URL: <u>https://zoom.us/webinar/register/WN_t6D6to08Sh2BhyoD3iw1HQ</u>



Date: November 22, 2022 Speaker: Dr. Travis Legleiter Title: Implementing Defensive Shifts Against Problematic Kentucky Weeds Registration URL: https://zoom.us/webinar/register/WN_QnugWPJJQUynBXDf4io9zg



Date: December 6, 2022 Speaker: Dr. Kiersten A. Wise Title: Corn Disease Management Questions Asked in 2022 Registration URL: <u>https://zoom.us/webinar/register/WN_KwibLTsHQY6oJjiKzURCEQ</u>



Date: December 13, 2022 Speaker: Dr. Raul Villanueva Title: Entomological Studies in Corn and Soybeans Under Difficult Circumstances (Covid, a Tornado and Drought in 2022)

Registration URL: https://zoom.us/webinar/register/WN_3KVwBMYKQYKnxzW1K-A0-g



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SAVE THE DATE



2023 Kentucky Crop Health Conference

Agricultural Pest Management Conference @ The National Corvette Museum 350 Corvette Dr. • Bowling Green, Ky. 42101



February 9, 2023 9 a.m. to 5 p.m. CST

REGISTRATION IS NOW OPEN Limited to 100 participants https://KentuckyCropHealthConference2023.eventbrite.com

Lunch included CCA, Kentucky pesticide applicator credits available

An in-depth and enhanced conference covering insect pests, plant pathogens, and weeds affecting Kentucky corn, soybean, and wheat acres. Presentations by University of Kentucky Extension Specialists and invited nationally prominent Extension Specialists from across the United States.

Scan to register









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UPCOMING EVENTS

<u>UK 2022 Crop Pest Management Webinar Series</u> Managing Important Soilborne Diseases of Soybean
<u>UK 2022 Crop Pest Management Webinar Series</u> Implementing Defensive Shifts Against Problematic Kentucky Weeds
<u>UK 2022 Crop Pest Management Webinar Series</u> Corn Disease Management Questions Asked in 2022
UK 2022 Crop Pest Management Webinar Series Entomological Studies in Corn & Soybeans Under Difficult Circumstances (Covid, a Tornado & Drought) in 2022
UK Winter Wheat Meeting
KY Commodity Conference - Bowling Green
KATS In-depth Mode of Action
KATS Soil Fertility and Assessment
National Commodity Classic - Orlando FL
UK Wheat Field Day
KATS Crop Scouting Clinic
KATS Drone Pilot Certification Prep Course
KATS Spray Clinic
UK Corn, Soybean and Tobacco Field Day

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